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SYNTHESIS OF ACETYLATED GLYCOSIDES OF

HYDROXYNAPHTHOQUINONES

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A method is proposed for the synthesis of acetylated glycosides of hydroxynaphthoquinones. The condensation of D-glucose and D-galactose (tert-butyl orthoacetate)s with lawsone and lapachol has given the tetra-O-acetyl- β -D-glucopyranosides of lawsone and of lapachol and the tetra-O-acetyl- β -galactopyranoside of lawsone. The structures of the glycosides obtained have been confirmed by IR and ¹H and ¹³C NMR spectroscopy. The structure of the lawsone acetylgalactopyranoside described previously has been corrected.

The majority of investigation of recent years in the field of the synthesis of glycosides of hydroxynaphthoquinones have been connected with the creation of water-soluble hydroxynaphthoquinone derivatives, which are necessary for studying the influence of the carbohydrate moieties on their biological activity. The tetra-Oacetyl- β -D-glucopyranosides of lawsone (40%) and menoctone (56%) with the o-quinoid structure of the aglycone [1], and the tetra-O-acetyl-D-glucopyranosides of lawsone (30%) and of lapachol (16%) and the tetra-Oacetyl-D-galactopyranoside of lawsone (28%) with the p-quinoid structure of the aglycone [3] have been obtained by using various modifications of the Koenigs-Knorr method. The acetates of glycosides of lawsone and lapachol exhibited antitumoral activity [2, 3]. The reduction of an acetylated D-glucose residue leads to a marked increase in the immunodepressive action of lawsone [4].

The biological activity of glycosides of hydroxynaphthoquinones makes necessary a search for new and more effective methods for their synthesis.

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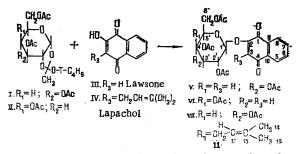
C atom	Compound				
	111†	v	VI	VIII	VII
1 2 3 4 5 6 7 8 9 10 11 12 13 14	180 1 160,2 109,8 184,8 126,6 134,3 133,1 126,1 131,0 132,0	178 8 157.4 114,9 184.8 126.7 134.2 133.6 126.1 131.1 131.7	178 8 157.4 114.8 184.4 126.7 134.3 133.7 126,1 131.0 131.6	179 2 157.1 115.4 184.8 126.7 134.3 133.6 126.2 131.2 131.8	180,9 152,6 137 8 184,8 126,5 134,1 126,0 131,2 131,9 25,8 119,4 134,1 18,1
15 1' 2' 3' 4' 5' CH ³ COO CH ³ COO		97.8 70.6 72.3 4 68,1 72 9 4 61.9 20.5	98,3 71,9 ‡ 70,5 ‡ 67,9 ‡ 66,9 ‡ 61,6 20,5 170	95.1 68.4‡ 67.5‡ 67.5‡ 61.4 20.5 170	23,5 99,2 71,7 72,2 68,4 72,7 61,7 20,5 170

TABLE 1. ¹³C Chemical Shifts of Hydroxynaphthoquinone Glycoside (ppm relative to TMS)*

* The assignment of the signals from the 13 C spectra of the quinoid part of compounds (III, V-VII) was made on the basis of [10] and the assignment of the signals in the carbohydrate moieties of glycosides (V-VIII) on the basis of [11]. † Methyl ester.

‡Assignment ambiguous.

Basing ourselves on a method described in the literature [5, 6], we have proposed a convenient method for the glycosylation of lawsone (III) [7], which consists in boiling equimolar amounts of the quinone with orthoesters of glucose (I) and of galactose (II) in absolute chlorobenzene. The proposed method of glycosylating hydroxy-1,4-naphthoguinones is stereospecific and gives the β -D-glycosides (V-VII) in good yields. An inert medium and the absence of any external catalysts whatever permits this method to be used for the glycosylation of labile hydroxynaphthoguinones.



The structures of glycosides (V-VII) were confirmed by the results of IR and NMR spectroscopy, and also by elementary analysis. Because of the tautomeric nature of the hydroxynaphthoquinones (III) and (IV), the formation of glycosides with either the o- or the p-quinoid structure of the aglycone is possible, in principle. The choice in favor of the p-quinoid structure of glycosides (V)-(VII) was made on the basis of a comparison of the ¹³C spectra of 2-methoxy-1,4-naphthoquinone and of compounds (V-VII) (Table 1). The values of the SSCCs $J_{1',2'} = 7.3$ Hz in the ¹H spectrum of compound (VII) shows the β configuration of the glycosidic bond. In the case of compounds (V) and (VI), in the ¹H spectra of which no isolated signals of anomeric protons were observed, the configuration of the glycosidic bond was established from the chemical shift of the anomeric carbon atom in the ¹³C spectra.

The conclusion of the β configuration of the glycosidic bond in (V) and (VI) was made on the basis of a comparison of the ¹³C spectra of (V) and (VI) and those of the β - and α -tetra-O-acetyl-D-galactopyranosides (VI) and (VIII) (see Table 1), which agrees well with the results of the work of Steinerova et al. [4].

The process of glycosylation of (III) and (IV) by the orthoesters (I) and (II) apparently takes place by a mechanism of protonic catalysis [8], since 2-hydroxy-1,4-naphthoquinones are medium-strength acids (for lawsone, pK = 4) [9]. This case is obviously an example of autocatalysis when the glycosylated alcohol is simultaneously an acid catalyst. It is interesting that 5-hydroxy-1,4-naphthoquinone, the proton of the hydroxy group of which is bound by a strong intramolecular hydrogen bond and is deprived of mobility to a considerable degree as compared with the analogous proton of lawsone was practically inert under the conditions of the proposed method of glycosylation.

When the physicochemical characteristics of the glycosides (V) and (VI) which we have synthesized were compared with those of analogous substances described previously [3], considerable discrepancies in the melting points were observed. A careful analysis of the ¹H and ¹³C spectra and a mixed melting point test of a sample of the galactopyranoside (VI) and compound (VIII) kindly supplied by Prof. M. M. de Oliveria showed that (VIII) has the α , not the β , configuration of the glycosidic bond, and the melting point of the glycoside (VI) that we had obtained did not agree with the literature figure [3]. Attempts to obtain lawsone glycosides by the method of [3] led to (V) and (VI) with the β configuration of the glycosidic bonds.

EXPERIMENTAL

Melting points were determined on a Boëtius stage. The specific rotations were measured on a Perkin-Elmer 141 polarimeter. NMR spectra were obtained on a Bruker HX-90 spectrometer with a working frequency of 90.0 MHz for ¹H and 22.6 MHz for ¹³C at 30°C in CDCl₃, with TMS as internal standard. IR spectra were recorded on a Specord-IR spectrophotometer in CHCl₃. TLC was performed on Silufol plates (Czechoslovakia) in the hexane-benzene-acetone (2:1:1) system. The plates were previously saturated with NH₃ vapors to prevent decomposition of the orthoester during the process of chromatography. Spots were revealed by heating the plates. Column chromatography was performed on SiO₂ L (40-60 u) (Czechoslovakia) in the hexane-acetone (10:1-2:1) system. The results of the elementary analyses of all the newly obtained compounds were in satisfactory agreement with the calculated values.

2-Hydroxy-1,4-naphthoquinone (lawsone) (III) was obtained as described by Donaldson [12], lapachol (IV) from lawsone as described by Jacobsen and Torrssel [13], 3,4,6-tri-O-acetyl- α -D-glucopyranosyl 1,2-(tert-butyl orthoacetate) (I) according to [14], and 3,4,6-tri-O-acetyl- α -D-galactopyranosyl 1,2-(tert-butyl orthoacetate) (II) according to [15].

<u>General Procedure for the Glycosylation of Hydroxynaphthoquinones.</u> A mixture of an orthoester (1 mmole), a hydroxynaphthoquinone (1 mmole), and 15 ml of absolute chlorobenzene was boiled until the orthoester had disappeared completely from the reaction mixture (1-6 h). The solvent was evaporated off in vacuum, the residue was dissolved in 50 ml of CHCl₃, and the solution was washed successively with 1 M K₂CO₃ solution (2×20 ml) and with water (2×20 ml) and was dried over Na₂SO₄. The solvent was eliminated and the desired glycoside was isolated from the residue by crystallization or by column chromatography. The combined wash-waters were acidified with HCl, extracted with ethyl acetate (2×20 ml), washed with water (2×20 ml), and dried with Na₂SO₄. The solvent was driven off, and the residue consisted of the chromatographically pure quinone.

 $\underline{2-(\text{Tetra-O-acetyl-}\beta-\text{D-glucopyranosyloxy})-1,4-\text{naphthoquinone (V). Yield 64\%. C_{24}H_{24}O_{12} \cdot 0.5 \text{ CH}_{3}OH. } \\ \text{mp 166.5-168.5°C (MeOH). } [\alpha]_{D}^{22} -30.6 (c1.0; CHCl_3). \text{ Lit. [4], mp 168.5-170°C (ethanol). }^{1}H \text{ spectrum (}^{\delta}, \text{ppm):} \\ 2.05-2.12 (m, 12 \text{ H}, 4 \times OAc); 3.90-4.01 (m, 1 \text{ H}, \text{H}_{5}^{1}); 4.21-4.25 (m, 2 \text{ H} 2\text{H}_{6}^{1}); 5.10-5.42 (m, 4 \text{ H}, \text{H}_{1}^{1}, \text{H}_{2}^{1}, \\ \text{H}_{3}^{1}, \text{H}_{4}^{1}); 6.41 (s, 1 \text{ H}, \text{H}_{3}); 7.61-7.84 (m, 2 \text{ H}, \text{H}_{6}, \text{H}_{7}); 8.02-8.15 (m, 2 \text{ H}, \text{H}_{5}, \text{H}_{8}). \text{ IR spectrum (}\nu \text{ cm}^{-1}): 1758 (CH_{3}COO), 1689 \text{ and 1658 (C=O).} \\ \end{array}$

 $\frac{2-(\text{Tetra-O-acetyl}-\beta-\text{D-galactopyranosyloxy})-1,4-\text{naphthoquinone (VI)}. \text{ Yield 70\%. C}_{24}\text{H}_{24}\text{O}_{12}. \text{ mp 176-178 °C (MeOH), } [\alpha]_{D}^{22} -16.6^{\circ} \text{ (c 1.0; CHCl}_{3}\text{)}. \text{ }^{1}\text{H spectra } \delta, \text{ ppm}\text{)}: 2.01-2.19 \text{ (m, 12 H, } 4\times \text{ OAc}\text{)}; 4.18 \text{ (m, 3 H, } \text{H}_{5}^{\circ}, 2 \text{ H}_{6}^{\circ}\text{)}; 5.12-5.22 \text{ (m, 2 H, H}_{2}^{\circ}, \text{H}_{3}^{\circ}\text{)}; 5.47-5.54 \text{ (m, 2 H, H}_{1}^{\circ}, \text{H}_{4}^{\circ}\text{)}; 6.46 \text{ (s, 1 H, H}_{3}\text{)}; 7.69-7.79 \text{ (m, 2 H, H}_{6}^{\circ}\text{H}_{7}^{\circ}\text{)}; 8.02-8.06 \text{ (m, 2 H, H}_{5}, \text{H}_{8}\text{)}. \text{ IR spectra } (\nu, \text{ cm}^{-1}): 1758 \text{ (CH}_{3}\text{COO}\text{)}, 1686 \text{ and } 1656 \text{ (C=O)}.$

 $\frac{2-(\text{Tetra-O-acetyl}-\beta-\text{D-glucopyranosyloxy})-3-(3-\text{methylbut}-2-\text{enyl})-1,4-\text{naphthoquinone (VII)}. \text{ Yield} \\ 72\%. \text{ mp 62-63°C (MeOH). Lit. [2]: mp 62-65°C (MeOH-ethyl acetate). }^{\text{H}} \text{ spectrum } (\delta, \text{ppm}): 1.66 \text{ (s, 3 H, CH_3)}; 1.79 \text{ (s, 3 H, CH_3)}; 2.00-2.12 \text{ (m, 12 H, } 4\times\text{OAc}), 2.33 \text{ (d, J} = 6.7 \text{ Hz}, 2 \text{ H}, 2 \text{ H}_{11}); 3.90-4.00 \text{ (m, 1 H, H}_{5}'); 4.08-4.20 \text{ (m, 2 H, 2 H}_{6}'); 5.10-5.33 \text{ (m, 4 H, H}_{12}, \text{H}_{2}', \text{H}_{3}', \text{H}_{4}'); 5.82 \text{ (d, } J_{1',2'} = 7.3 \text{ Hz}, 1 \text{ H}, 1 \text{ H}_{1}'); 7.65-7.75 \text{ (m, 2 H, H}_{6}, \text{H}_{7}); 8.00-8.09 \text{ (m, 2 H, H}_{5}, \text{H}_{8}). \text{ IR spectrum } (\nu, \text{ cm}^{-1}: 1758 \text{ (CH}_{3}\text{COO}), 1673 \text{ and } 1623 \text{ (C=O)}. \end{cases}$

 $\frac{2-(\text{Tetra-O-acetyl-}\alpha-\text{D-galactopyranosyloxy})-1,4-\text{naphthoquinone (VIII).}}{(VIII).} \text{ mp } 210-211^{\circ}\text{C} (\text{MeOH-ethyl acetate}). [\alpha]_{D}^{22} -170.4^{\circ} (c: 0.5 \text{ CHCl}_{3}).$ ¹H spectrum (δ , ppm): 1.98-2.18 (m, 12 H, 4× OAc); 4.08-4.23 (m, 3 H,

H₅, 2 H₆'; 5.35-5.38 (m, 1 H, H₃); 5.56-5.65 (m, 2 H, H₂', H₄'); 5.88 (d, $J_{1^{1},2^{1}} = 3.2 \text{ Hz}, 1 \text{ H}, 1 \text{ H}_{1}^{1}$); 6.54 (s, 1 H, H₃); 7.70-7.80 (m, 2 H, H₆, H₇); 8.04-8.12 (m, 2 H, H₅, H₈). IR spectrum (ν , cm⁻¹): 1748 (CH₃COO), 1681 and 1648 (C=O).

SUMMARY

A method is proposed for the synthesis of acetylated glycosides of hydroxynaphthoquinones.

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SYNTHESIS OF GLUCOSIDES OF 3-ALK[EN]YL-

2-HYDROXY-1,4-NAPHTHOQUINONES

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The condensation of D-glucose (tert-butyl orthoacetate) with 3-alk[en]yl-2-hydroxy-1,4naphthoquinones has yielded a series of acetylated glycosides of hydroxynaphthoquinones. It has been established that the time of the glycosylation reaction lengthens with an increase in the length and in the degree of branching of the side chain of the quinone. It has been shown that when the glycosides obtained are deacetylated cleavage of the glycosidic bond takes place with the formation of glucose and the corresponding quinone methyl ethers. Details of IR and ¹H and ¹³C NMR spectra are given.

We have previously [1, 2] reported a new method of obtaining acetylated glycosides of hydroxynaphthoquinones which consists in boiling equimolar amounts of a quinone and an orthoester in absolute chlorobenzene without a catalyst. To elucidate the possibilities of the proposed method of glycosylation, a number of 3-alk-[en]yl-2-hydroxy-1,4-naphthoquinones have been condensed with the glucose orthoesters (I) and (II). The results are given in Table 1.

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